













Rusting & Corrosion

Rusting is the **corrosion** of **iron** (& steel).



Other metals can corrode but not rust





Corrosion

Corrosion is also the reaction of a metal with **oxygen**.

It is sometimes called **oxidation**, and can happen to most **metallic** elements.

Metal + Oxygen \rightarrow Metal oxide



Rusting

Rusting is corrosion of a metal containing **iron**. It is also sometimes called an **oxidation**.

BUT...both water and oxygen (from the air) are required for rusting

Iron + Oxygen + Water \rightarrow Hydrated Iron Oxide (Rust)

Rusting



Salt and other electrolytes (e.g. **acid**) can speed up rusting/corrosion

Rusting & Corrosion

Corrosion is when a ______ element reacts with **oxygen** to form a metal ______. **Rusting** is a special form of ______ that happens when **iron** (or **alloys** that contain **iron**) react with and _____. Both **rusting** and **corrosion** can be described as _____. Salt and other electrolytes (for example, acids) do not cause **rusting/corrosion** on their own, but they can cause it to if **oxygen** and **water** are present.

Metallic	oxide	corrosion	oxygen	water	oxidations
speed up					

Group practical:

Design an experiment to prove that **oxygen** and **water** are needed for **rusting** to occur and that **salt** increases the rate of **rusting**

Work in groups of 3-4 (on your double tables)

Group practical:

A basic equipment set up...



Group practical:

How could we test if rusting happens **without oxygen**?

How could we test if rusting happens **without** water?

HINTS:

Boiled water doesn't have any **oxygen** in it... **Oil** floats on water and **doesn't let air through it**...

We have **silica**, which can **remove water** from the **air**...

